

Proposal to Perform In-Situ Chemical Oxidation for the Treatment of Source and Residual Contamination via Ferric Oxide Activated Persulfate and Catalyzed Hydroxyl

То

Tetra Tech

For

Hoff VC HSCA Site 334 Layfield Road New Hanover Township, PA

March 2022

Innovative Environmental Technologies, Inc. 6071 Easton Road Pipersville, PA 18947 (888) 721-8283 www.IET-INC.net February 16th, 2022

Johnathan Dziekan Senior Project Manager Tetra Tech, Inc.

Dear Mr. Dziekan,

Innovative Environmental Technologies Inc. (IET) has completed a remedial design and quotation for the remediation of the delineated contamination at the site, Hoff Vinyl Chloride (VC) HSCA Site, at 334 Layfield Road in New Hanover Township, Pennsylvania.

The contaminants of concern at the subject site are Volatile Organic Compounds (VOC's) and 1,4-Dioxane. As a result of IET's evaluation of the provided analytical data, monitoring well logs and data collected, IET is pleased to provide a quote utilizing an activated persulfate and catalyzed hydroxyl technology. The following proposal offers the price to implement chemical oxidation with both Sodium Persulfate activated by Ferric Oxide, and Hydrogen Peroxide activated by ferrous sulfate. The remedial design is presented as one treatment area with fourteen permanent injection well locations (IW-1-IW-6 A/B and MW-F A/B). The lump sum cost for the proposed design is IET's United States Apparatus Patent Number 7,044,152.

The following proposal will set-forth a lump sum price for the implementation and follow up of the remedial process. All costs included in the lump sum price are listed below.

- All chemicals and materials necessary to complete the proposed plan
- All equipment and personnel required to execute the proposed plan
- Handling and Management of materials on site
- Mobilization/Demobilization of the required crews
- All per diem for the required crews
- Health and Safety Plan for the site
- Site Restoration
- Final field injection report
- Final plot of injection points
- Six data analysis reports, based on data provided by Tetra Tech

Table of Contents

OBJECTIVE	3
TREATMENT AREA	3
TECHNOLOGY DISCUSSION	4
SCOPE OF WORK	6
SUMMARY	11
APPENDIX 1: SITE MAP	12
APPENDIX 2: DOSAGE CALCULATIONS	12
PERMANENT WELL CALCULATIONS	13
INJECTION WELL INSTALLATION	20

OBJECTIVE

It shall be the objective of IET to conduct a chemical oxidation event at the site located in New Hanover, PA. A unique ISCO process will be implemented in order to directly oxidize the contaminants of concern and stimulate a long lasting in-situ bioremediation process. Multiple injection wells are proposed to treat residual contamination present. The proposed and existing wells are pictured below.



TREATMENT AREA

The defined treatment area will target a 9,076 square foot area and will require 14 permanent injection well points. A 17' radius of influence is estimated across the vertical interval targeted at each injection location. The ROI is estimated based on the site's available geologic data, utilizing historic boring logs to estimate a soil porosity along with assumptions based off of pre vs. post injection groundwater data collected at other sites with similar geology. The soil Freundlich absorbsion correction is assumed moderate in order account for varied VOC's targeted and their varied partitioning characteristics to soil (IET has assumed the value to be 5%). The Freundlich equation is an adsorption isotherm that relates the concentration of a solute on the surface of an adsorbent to the concentration of the solute in a liquid. The Freundlich equation is used to determine the theoretical mass of contamination adsorbed to the soil. The mass of contaminant in the soil was determined using the soil adsorption correction (item 1). The K constant is a figure relating the capacity of the adsorbent for an adsorbate and the 1/n constant is a

function of the strength of adsorption (*American Water Works Association*, Water Quality and Treatment, 1999). The Freundlich equation is listed below:

 $q_e = KC_e^{1/n}$

The theoretical values of K and 1/n are found in the following references: (Dobbs and Cohen, 1980/Faust and Aly, 1983).

Treatment area calculations are located below in Appendix 2.

IET estimates that this injection event will take 7 day(s) to implement. The price present herein is guaranteed *regardless* of the actual field time required to implement. The lump sum pricing assumes a onsite source of water and secure storage of the amendment.

TECHNOLOGY DISCUSSION

Advanced Oxidation

Oxidation is defined as a chemical process in which electrons are transferred from an atom, ion or compound. The *in-situ* chemical oxidation process is designed to destroy organic contaminants either dissolved in groundwater, sorbed to the aquifer material, or present as free product. Oxidants most frequently used in chemical oxidation include hydrogen peroxide (H_2O_2) , potassium permanganate $(KMnO_4)$, persulfate $(Na_2O_8S_2)$ and ozone (O_3) . Peroxone, which is a combination of ozone and hydrogen peroxide, is also used. Fenton's Reagent, which is hydrogen peroxide mixed with a metal catalyst, commonly an iron catalyst, can also be used. *In-situ* chemical oxidation (ISCO) can be accomplished by introducing chemical oxidants into the soil or aquifer at a contaminated site using a variety of injection and mixing apparatuses. Normally, vertical or horizontal injection wells are used to deliver chemical oxidants. *Ex-situ* oxidation is accomplished by pumping groundwater from extraction wells and treating the groundwater above ground. In the recirculation approach, oxidants can be mixed with the extracted groundwater, which is subsequently pumped back into the aquifer through injection wells.

What are the advantages and disadvantages of chemical oxidation?

Chemical oxidation offers several advantages over other *in-situ* or *ex-situ* remediation technologies for petroleum compounds:

• The greatest advantages are the rapid treatment time and the ability to treat contaminants present at high concentrations.

• It is effective on a diverse group of contaminants and can often achieve maximum clean-up results.

What contaminants can be treated with chemical oxidation?

Common contaminants treated by chemical oxidation are amines, phenols, chlorophenols, cyanides, halogenated aliphatic compounds, mercaptans, BTEX compounds, MTBE and certain pesticides in liquid waste streams. Oxidation effectiveness depends on the organic compound.

Is chemical oxidation safe?

While the use of chemical oxidation can be quite safe if done properly, there are significant potential hazards. Most oxidants are corrosive. This means that they have the ability to burn the skin and wear away certain materials. Chemical oxidation also has some disadvantages. The disadvantages are as follows:

• Oxidation is nonselective. As such, the oxidant will not only react with the target contaminants but also with substances found in the soil that can be readily oxidized. In the case of gaseous ozone, the ozone can react with water and decompose to oxygen. Oxygen production can lead to serious problems such as the development of high pressures below the ground surface and possible explosions.

• Control of pH, temperature, and contact time is important to ensure the desired extent of oxidation.

How long does chemical oxidation take?

The time required to clean up a contaminated site using chemical oxidation is dependent on the reactivity of the contaminant with the oxidant, the size and depth of the contaminated zone, the speed and direction of groundwater flow and type of soils and the conditions present at the contaminated facility. Generally, chemical oxidation is more rapid than other treatment technologies. The time scale is usually measured in months, rather than years.

In-situ oxidation uses contact chemistry of the oxidizing agent to react with volatile organic compounds, munitions, certain pesticides and wood preservatives. The most common oxidizers used in soil and groundwater remediation are hydrogen peroxide (and the hydroxyl radical), potassium permanganate, and ozone, which are non-selective oxidizers. Other oxidants are available, but are used less due to cost, time or potential toxic by-products.

Technology Selection

Persulfate is activated by Fe III which requires a lower activation energy than alternative mechanisms while not consuming the persulfate oxidant. The chosen activation mechanism is believed to elevate the oxidation state of the iron transiently to a supercharged iron ion which in itself may act as an oxidant. As this supercharged iron cation is consumed, the resulting ferric species can act as a terminal electron acceptor for biological attenuation. Coincidentally, the generated sulfate ion from the decomposition of the persulfate provides a terminal electron acceptor for sulfate reducers which may further remediate the targeted compounds in the groundwater and soils. In addition, for the injection locations where the primary compound of concern is 1,4 dioxane catalyzed hydroxyl radicals are proposed to be injected with 2-5% hydrogen peroxide and ferrous sulfate as the injectant in order to re-create oxidizing conditions in the groundwater near these locations. The desired ISCO reactions that occur in the subsurface include persulfate radicals and ferrate, as summarized below (Equation 1):

 $S_2O_8^{-2} + Fe^{+3} - Fe^{(+4 \text{ to}+6)} + SO_4^{2-} + SO_4^{2-}$ (Eq. 1)

Secondary Attenuation Processes

After dissolved oxygen has been depleted in the treatment area, sulfate (a by-product of the persulfate oxidation) may be used as an electron acceptor for anaerobic biodegradation by indigenous microbes. This process is termed sulfidogenesis and results in the production of sulfide. Stoichiometrically, each 1.0 mg/L of sulfate consumed by microbe's results in the destruction of approximately 0.21 mg/L of BTEX compounds. Sulfate can play an important role in bioremediation of petroleum products, acting as an electron acceptor in co-metabolic processes as well. For example, the basic reactions for the mineralization of benzene and toluene under sulfate reducing conditions are presented in equations 2 and 3:

$$C_6H_6 + 3.75 \text{ SO}_4^{-2-} + 3 \text{ H}_2\text{O} \longrightarrow 0.37 \text{ H}^+ + 6 \text{ HCO}_3^{--} + 1.87 \text{ HS}^- + 1.88 \text{ H}_2\text{S}^-$$
 (Eq. 2)
 $C_7H_8 + 4.5 \text{ SO}_4^{-2-} + 3 \text{ H}_2\text{O} \longrightarrow 0.25 \text{ H}^+ + 7 \text{ HCO}_3^{--} + 2.25 \text{ HS}^- + 2.25 \text{ H}_2\text{S}^-$ (Eq. 3)

Ferric iron is also used as an electron acceptor during anaerobic biodegradation of many contaminants, sometimes in conjunction with sulfate. During this process, ferric iron is reduced to ferrous iron, which is soluble in water. Hence, ferrous iron may be used as an indicator of anaerobic activity. As an example, Stoichiometrically, the degradation of 1 mg/L of BTEX results in the average consumption of approximately 22 mg/L of ferric iron (or "production" of ferrous iron) as shown below (equations 4-6).

 $C_6H_6 + 18 H_2O + 30 Fe^{3+} -----> 6 HCO_3 - + 30 Fe^{2+} + 36 H^+ (Eq. 4)$ $C_7H_8 + 21 H_2O + 36 Fe^{3+} ----> 7 HCO_3 - + 36 Fe^{2+} + 43 H^+ (Eq. 5)$ $C_8H_{10} + 24 H_2O + 42 Fe^{3+} ----> 8 HCO_3 - + 42 Fe^{2+} + 50 H^+ (Eq. 6)$

While ferrous iron is formed as a result of the use of the ferric species as a terminal electron acceptor, residual sulfate is utilized as a terminal electron acceptor by facultative organisms thereby generating sulfide under these same conditions. Together, the ferrous iron and the sulfide promote the formation of pyrite as a remedial byproduct (equation 7). This reaction combats the toxic effects of sulfide and hydrogen sulfide accumulation on the facultative bacteria, while also providing a means of removing targeted organic and inorganic COIs via precipitation reactions. Moreover, pyrite possesses a high number of reactive sites that are directly proportional to both its reductive capacity and the rate of decay for the target organics.

Fe²⁺ + 2S²⁻ -----> FeS₂ + 2e (Eq. 7)

SCOPE OF WORK

Subsurface Pathway Development

Initially, compressed air shall be delivered to the subsurface via IET proprietary injection trailer system. This process step allows for confirmation of open delivery routes while enhancing horizontal injection pathways. The confirmation of open and viable subsurface delivery pathways insures that upon introduction of the oxidizer(s) injections will occur freely thus minimizing health and safety risks associated with oxidant full injection lines and injection tooling when no subsurface delivery route has been established. Confirmation of open and free pathways is accomplished via observed pressure drops and fee moving compressed gases to the subsurface.

Oxidant Injection

A colloidal suspension of the ferric-based catalysed persulfate is immediately injected into the subsurface pathways and voids that were developed during the compressed air injection step, under constant pressure ranging from 10-110 psi. A small amount of water follows this step in order to rinse the injection equipment. IET expects the need of the liquid pressures to fall in the range of **30-75** psi in order to introduce the material into the lithology documented onsite.

Post Liquid Injection – Compressed Air Injection

Lastly, the injection lines are cleared of liquids and all injectants are forced into the created formation and upward into the vadose zone. This step insures that all material is injected outward into the formation and minimizes any surface excursions of injectants following the release of the injection pressure. Once the injection cycle is complete, the injection point is temporarily capped to allow for the pressurized subsurface to accept the injectants.

Permanent Injection Well Injection

Similar procedure is followed when injecting into permanent injection wells, fourteen existing wells are proposed to be injected at the Hoff site, a pvc fitting will be attached to the riser and injection will be attempted across the entire well's screen. Three additional Injection wells are proposed to be installed (IW-4-IW-6) these are proposed to be screened at two isolated depths from 30-35' and 20-25'. A Construction diagram of the injection wells is included below. The new wells will be installed using 4.25" HSA and 3.75" air rotary to depth once HSA refusal is encountered. Pricing for installation of two additional monitoring wells is also included in this proposal.



Equipment Description

The injections shall occur via IET's mobile injection trailer and permanent injection wells as described. The injection lines are composed of the following: stainless steel fittings, Viton seals, chemical resistant one inch and ¾ inch diameter hose. The patented injection system includes: two 200 gallon conical tanks capable of maintaining 30% solids as a suspension via lightning mixers; on-board generator, all stainless steel tig welded piping, 2" pneumatic diaphragm pump with an operating pressure of 110 psi.; on-board 25 CFM/175 psig compressor with 120 gallons of air storage; self-contained eye wash and safety shower. IET proprietary injection rods with retractable injection zones and backflow protection Injection zones of 18 inches are to be used in combination with 24" injection AWJ-Rods where appropriate.



Well Installation Equipment – AMS 9500 Series Track Rig

IET plans to use a 9500 Series AMS DPT rig to io install the permanent injection wells. This unit can be viewed below.



IET INJECTION SYSTEM UNITED STATES PATENT 7,044,152



Injection Trailers Include: Multiple Liquid Feed Systems, Stainless Steel Piping, Isolated Compressed Gas Containment, Safety Shower, Eyewash Station, Onboard Generator, Chemical Resistant Construction, Mobile Office Space



SUMMARY

Innovative Environmental Technologies, Inc. presents this proposal with one treatment area. It is estimated to cost . It is assumed that it will take 7.0 days to complete the remedial program. The lump sum price is independent of the amount of time that the injection event takes.

	Area (FtXFt)	Mass of Soil/BR (tons)	Lbs FeO/Persulfate	Gallons 50% H2O2	Lbs FeSO4	Injection Pts	Days
Treatment Areas							
IW-A	907	403.32	0	12	23	1	0.2
IW-B	907	403.32	0	18	23	1	0.2
IW-2A	907	403.32	0	26	23	1	0.2
IW-3A	907	403.32	0	26	23	1	0.2
IW-3B	908	403.32	0	26	23	1	0.2
IW-2b	908	403.32	770	0	0	1	0.25
New Wells	3,630	3,226.67	4,070	0	0	8	2.75
Total	9,076	5,646.56	4,840	107	115	14	4
				Line Items			Unit
				Well Installation			LS
				Materials Handling and	Forklift Rental		1
				Hazmat Shipping			LS
				Per Diem (2 man Crew)		7
				Private Utility Markout			0.5
				Administrative and Rep	orting Cost		LS
				Total Project Cost			

All Prices Quoted are Valid for 60 Days from the Date on Proposal

APPENDICES

APPENDIX 1: SITE MAP



DPT Injection Point

PNT	HOFF
401 402	NORTHING 370531.1550 370531.2120
403	370410.4810 [°] 370410.5280 [°]
405	370417,8980° 370417,9810°
407	370450.6210 370450.6690
410	370483.7820

APPENDIX 2: DOSAGE CALCULATIONS

PERMANENT WELL CALCULATIONS

Tetratech-Hoff VC, PA					
Pilot Area		Saturated Zone			
IW-A					
Parameters	Units	Assumptions			
Target Area	Ft.X Ft.	907.46			
Injection Radii	Ft	17			
Soil Absorbsion Correction for GAC Constant	%	5			
Area of influence of Remediation Injection(s)	Sq. Ft.	907.91951			
Estimated Number of Injections to Treat Area	# Injections	1			
vertical impacted zone	Ft.	10			
Target Zone		15-25			
Total Volume Targeted	Cu. Yd.	336.0962963			
Porosity	%	20.00%			
Mass of soil to be targeted	lbs	806631.1111			
Mass of soil to be targeted	grams	366210524.4			
Volume of Groundw ater targeted	gals	13611.9			
Contaminant Conc.	ppm	0.06	Calculations Targeted Compounds		
Mass of Contaminant - water	lb.	0.006819562	Ave Mol Mass of Targeted Compounds g/mol	95	
Mass of Contaminant -w ater	Grams	3.096081103	Moles	7.345584531	
Mass of Contaminant -soil	lb.	1.530252091	Mole Mass of H2O2	34	
Mass of Contaminant -soil	Grams	694.7344494	Moles of H2O2	820.9770947	
Mass of Contaminent Targeted	Grams	697.8305305	Mole Mass of Persulfate	238	
Mass of Contaminent Targeted	lbs	1.537071653	Moles of Na2S2O8	212.9702545	
Calculated soil conc.	ppm	1.905544718			
Grams of H2O2 Required	Grams	27913.22122			
Lbs of H2O2 Required	lbs	61.48286612			
Targeted percentage of FeSO4 in solution	%	0.02%	H2O2	111.76	
Pounds of FeSO4 required	pounds	23	Na2S2O8	28.99	
Injection Summary					
Number of Injection Locations		1.00			
Injection Depth		15-25'			
Pounds of FeSO4		23.00			
Galons of 50% H2O2		11.71			
Injection Point Summary - Number of Intervals		1.00			
Injection Zones	20-25'	1.00			
Pounds of FeSO4 per interval	20 20	23			
Gallons of 2% H2O2 per interval		293			
Cost Basis Per injection Event					
Cost Basis Event #1		\$/l Init			
Pounds of FeSO4 required	23.00	w on t			
Gallons of 50% H2O2 Required	12 00				
Number of Injection Points per Event	1.00				
Days of Injection Trailer and Rig	1.00				
Material Cost (Including on-site management etc)	0.2				
Total					
1 0100					

Tetratech-Hoff VC, PA					
Pilot Area		Saturated Zone			
IW-B					
Parameters	Units	Assumptions			
Target Area	Ft.X Ft.	907.46			
Injection Radii	Ft	17			
Soil Absorbsion Correction for GAC Constant	%	5			
Area of influence of Remediation Injection(s)	Sq. Ft.	907.91951			
Estimated Number of Injections to Treat Area	# Injections	1			
vertical impacted zone	Ft.	10			
Target Zone		25-35			
Total Volume Targeted	Cu. Yd.	336.0962963			
Porosity	%	20.00%			
Mass of soil to be targeted	lbs	806631.1111			
Mass of soil to be targeted	grams	366210524.4			
Volume of Groundwater targeted	gals	13611.9			
Contaminant Conc.	ppm	0.2	Calculations Targeted Compounds		
Mass of Contaminant - water	lb.	0.022731873	Ave Mol Mass of Targeted Compounds g/mol	95	
Mass of Contaminant -w ater	Grams	10.32027034	Moles	11.22741175	
Mass of Contaminant -soil	lb.	2.3266164	Mole Mass of H2O2	34	
Mass of Contaminant -soil	Grams	1056.283845	Moles of H2O2	1254.828372	
Mass of Contaminent Targeted	Grams	1066.604116	Mole Mass of Persulfate	238	
Mass of Contaminent Targeted	lbs	2.349348273	Moles of Na2S2O8	325.5159242	
Calculated soil conc.	ppm	2.912543591			
Grams of H2O2 Required	Grams	42664.16463			
Lbs of H2O2 Required	lbs	93.97393091			
Targeted percentage of FeSO4 in solution	%	0.02%	H2O2	111.76	
Pounds of FeSO4 required	pounds	23	Na2S2O8	28.99	
Injection Summary					
Number of Injection Locations		1.00			
Injection Depth		25-35'			
Pounds of FeSO4		23.00			
Galons of 50% H2O2		17.90			
Injection Point Summary - Number of Intervals		1.00			
Injection Zones	30-35'				
Pounds of FeSO4 per interval		23			
Gallons of 4% H2O2 per interval		224			
· · ·					
Cost Basis Per injection Event					
Cost Basis Event #1					
Pounds of FeSO4 required	23.00				T
Gallons of 50% H2O2 Required	18.00	1			Ť.
Number of Injection Points per Event	1.00	1			t
Days of Injection Trailer and Rig	0.2				t
Material Cost (Including on-site management etc)					
Total					Γ

Tetratech-Hoff VC, PA					
Pilot Area		Saturated Zone			
IW-2A					
Parameters	Units	Assumptions			
Target Area	Ft.X Ft.	907.46			
Injection Radii	Ft	17			
Soil Absorbsion Correction for GAC Constant	%	5			
Area of influence of Remediation Injection(s)	Sa. Ft.	907.91951			
Estimated Number of Injections to Treat Area	# Injections	1			
vertical impacted zone	Ft	10			
Target Zone		15-25			
Total Volume Targeted	Qu Yd	336 0962963			
Porosity	%	20.00%			
Mass of soil to be targeted	lbs	806631 1111			
Mass of soil to be targeted	grams	366210524.4			
Volume of Groundwater targeted	grand	13611.9			
Contaminant Conc	nom	0.55	Calculations Targeted Compounds		
Mass of Contaminant - water	b	0.062512651	Ave Mol Mass of Targeted Compounds g/mol	95	
Mass of Contaminant - water	Grams	28 38074344	Moles	16 10921574	
Mass of Contaminant - soil	lh	3 308358485	Mole Mass of H2O2	34	
Mass of Contaminant - soil	Grams	1501 994752	Moles of H2O2	1800 44176	
Mass of Contaminant Soli	Grams	1530 375496	Mole Mass of Persulfate	238	
Mass of Contaminent Targeted	lbe	3 370871136	Moles of Na2S208	467 0538820	
Calculated soil conc	nom	1 178950067	10163 01 1020200	407.0330023	
Grame of H2O2 Required	Grame	61215 01092			
Lbs of H2O2 Required	Grans	121 9249454			
Torgeted percentage of EcCOA in adultion	0/	0.020/	H2O3	111 76	
Payede of FacO4 assumed	%	0.02%	H202	111.70	
Founds of FeSO4 required	pounds	23	Na23206	20.99	
Injection Summary					
Number of Injection Locations		1.00			
Injection Depth		15-25'			
Pounds of FeSO4		23.00			
Galons of 50% H2O2		25.68			
Injection Point Summary - Number of Intervals		1.00			
Injection Zones	20-25				
Pounds of FeSO4 per interval		23			
Gallons of 5% H2O2 per interval		257			
Cost Basis Per injection Event					
Cost Basis Event #1					
Pounds of FeSQ4 required	23.00				
Gallons of 50% H2O2 Required	25.00				
Number of Injection Points per Event	1 00				
Days of Injection Trailer and Rig	1.00	, ,			
Material Cost (Including on-site management etc)	0.2				
Total					

Tetratech-Hoff VC, PA					
Pilot Area		Saturated Zone			
IW-3A					
Parameters	Units	Assumptions			
Target Area	Ft.X Ft.	907.46			
Injection Radii	Ft	17			
Soil Absorbsion Correction for GAC Constant	%	5			
Area of influence of Remediation Injection(s)	Sq. Ft.	907.91951			
Estimated Number of Injections to Treat Area	# Injections	1			
vertical impacted zone	Ft.	10			
Target Zone		15-25			
Total Volume Targeted	Cu. Yd.	336.0962963			
Porosity	%	20.00%			
Mass of soil to be targeted	lbs	806631.1111			
Mass of soil to be targeted	grams	366210524.4			
Volume of Groundwater targeted	gals	13611.9			
Contaminant Conc.	ppm	0.55	Calculations Targeted Compounds		
Mass of Contaminant - water	lb.	0.062512651	Ave Mol Mass of Targeted Compounds g/mol	95	
Mass of Contaminant -w ater	Grams	28.38074344	Moles	16.10921574	
Mass of Contaminant -soil	lb.	3.308358485	Mole Mass of H2O2	34	
Mass of Contaminant -soil	Grams	1501.994752	Moles of H2O2	1800.44176	
Mass of Contaminent Targeted	Grams	1530.375496	Mole Mass of Persulfate	238	
Mass of Contaminent Targeted	lbs	3.370871136	Moles of Na2S2O8	467.0538829	
Calculated soil conc.	mag	4.178950067			
Grams of H2O2 Required	Grams	61215.01983			
Lbs of H2O2 Required	lbs	134.8348454			
Targeted percentage of FeSO4 in solution	%	0.02%	H2O2	111.76	
Pounds of FeSQ4 required	pounds	23	Na2S2O8	28.99	
Injection Summary					
Number of Injection Locations		1.00			
Injection Depth		15-25'			
Pounds of FeSO4		23.00			
Galons of 50% H2O2		25.68			
Injection Point Summary - Number of Intervals		1.00			
Injection Zones	20-25				
Pounds of FeSO4 per interval		23			
Gallons of 5% H2O2 per interval		257			
Cost Basis Per injection Event					
Cost Basis Event #1					
Pounds of FeSO4 required	22.00				
Gallons of 50% H2O2 Required	23.00				
Number of Injection Points per Event	20.00				
Days of Injection Trailer and Rig	1.00	·			
Material Cost (Including on-site management etc)	0.2				-
Total					

Tetratech-Hoff VC, PA					
Pilot Area		Saturated Zone			
IW-3B					
Parameters	Units	Assumptions			
Target Area	Ft.X Ft.	907.46			
Injection Radii	Ft	17			
Soil Absorbsion Correction for GAC Constant	%	5			
Area of influence of Remediation Injection(s)	Sa. Ft.	907.91951			
Estimated Number of Injections to Treat Area	# Injections	1			
vertical impacted zone	Ft	10			
Target Zone		25-35			
Total Volume Targeted	Cu. Yd.	336.0962963			
Porosity	%	20.00%			
Mass of soil to be targeted	lbs	806631.1111			
Mass of soil to be targeted	grams	366210524.4			
Volume of Groundwater targeted	gals	13611.9			
Contaminant Conc.	ppm	0.55	Calculations Targeted Compounds		
Mass of Contaminant - water	lb.	0.062512651	Ave Mol Mass of Targeted Compounds g/mol	95	
Mass of Contaminant -w ater	Grams	28.38074344	Moles	16.10921574	
Mass of Contaminant - soil	lb	3 308358485	Mole Mass of H2O2	34	
Mass of Contaminant -soil	Grams	1501.994752	Moles of H2O2	1800.44176	
Mass of Contaminent Targeted	Grams	1530.375496	Mole Mass of Persulfate	238	
Mass of Contaminent Targeted	lbs	3.370871136	Moles of Na2S208	467.0538829	
Calculated soil conc	nom	4.178950067			
Grams of H2O2 Required	Grams	61215 01983			
Lbs of H2O2 Required	lbs	134.8348454			
Targeted percentage of FeSO4 in solution	%	0.02%	H2O2	111 76	
Pounds of FeSO4 required	nounds	23	Na2S2O8	28.99	
	poundo	20		20.00	
Injection Summary					
Number of Injection Locations		1.00			
Injection Depth		25-35'			
Pounds of FeSO4		23.00			
Galons of 50% H2O2		25.68			
Injection Point Summary - Number of Intervals		1.00			
Injection Zones	30-35'				
Pounds of FeSO4 per interval		23			
Gallons of 5% H2O2 per interval		257			
Cost Posis Parinisation Event					
Cost Basis Fer Injection Event					
Cost Basis Event #1					
Pounds of FeSO4 required	23.00	t.			
Gallons of 50% H2O2 Required	26.00	d .			
Number of Injection Points per Event	1.00	d .			
Days of Injection Trailer and Rig	0.2				
Material Cost (Including on-site management etc)					
Total					

Tetratech					
Hoff VC Site Montco PA		Saturated Zone			
IW-2b					
Parameters	Units	Assumptions			
Target Area	Ft.X Ft.	907.46			
Injection Radii	Ft	17			
Soil Absorbsion Correction for GAC Constant	%	5			
Area of influence of Remediation Injection(s)	Sq. Ft.	907.9			
Estimated Number of Injections to Treat Area	# Injections	1			
vertical impacted zone	Ft.	10			
Target Zone		25-35'			
Total Volume Targeted	Cu. Yd.	336			
Porosity	%	20.00%			
Mass of soil to be targeted	lbs	8.07E+05			
Mass of soil to be targeted	grams	3.66E+08			
Volume of Groundwater targeted	gals	1.36E+04			
		TVOCs			
Contaminant Conc.	ppm	30	Calculations Targeted Compounds		
Mass of Contaminant - water	lb.	3.4	Ave Mol Mass of Targeted Compounds g/mol	92	
Mass of Contaminant -w ater	Grams	1548.0	Moles	82.9	
Mass of Contaminant -soil	lb.	13.4	Mole Mass of H2O2	34	
Mass of Contaminant -soil	Grams	6081.6	Moles of H2O2	#REF!	
Mass of Contaminent Targeted	Grams	7629.6	Mole Mass of Persulfate	238	
Mass of Contaminent Targeted	lbs	16.8	Moles of Na2S2O8	1205.4	
Calculated soil conc.	ppm	20.8			
Ratio of S2O3 to targeted Compouns	Ratio	20			
Grams of sodium persulfate	grams	152591.9	Molar Ratio Calc		
Pounds of Sodium Persulfate Required	Pounds	336.1	Targeted Compounds	1	
Allocation per compound (persulfate)	%	100.0%			
Total Pounds of Sodium Persulfate Required	Pounds	336.1			
Decomposition Rate of Sodium Persulfate	%/dav	1.10%			
Targeted Longevity of Persulfate	davs	80			
Total Persulfate Calcualted dosage	pounds	632			
Total Pounds of Ferric Oxide Required	Pounds	63			
Injection Summary					
Number of Injection Locations		1			
Injection Depth		30-35'			
Pounds of Persulfate		700			
Pounds of Ferric Oxide		70			
Pounds of Persulfate/Ferric Oxide		770			
Minimum Pounds of MgO					
Actual Pounds of MgO		0			
Injection Point Summary - Number of Intervals		1			
Injection Zones		30-35'			
Pounds of Sodium Persulfate w / longevity	per well	700	Gallons of Water needed (area)	400	
Pounds of Ferric Oxide	per well	70			
Pounds of Persulfate/Ferric	per well	770			
Cost Basis Summary	Units				
Pounds of Persulfate/Ferric Oxide	770	1			
Number of Injection Points per Event	1	1			
Days of Injection Trailer (3 Man Crew)	0.25	1			
	0.20	1			
		1			

Tetratech				
Hoff VC Site Montco PA		Saturated Zone		
New Wells				
Parameters	Units	Assumptions		
Target Area	Ft.X Ft.	3630		
Injection Radii	Ft	17		
Soil Absorbsion Correction for GAC Constant	%	5		
Area of influence of Remediation Injection(s)	Sa. Ft.	907.9		
Estimated Number of Injections to Treat Area	# Injections	4		
vertical impacted zone	Ft.	20		
Target Zone		15-35		
Total Volume Targeted	Qu. Yd.	2689		
Porosity	%	20.00%		
Mass of soil to be targeted	lbs	6.45E+06		
Mass of soil to be targeted	grams	2 93E+09		
Volume of Groundwater targeted	granio	1.09E+05		
	gais	TVOCs		
Contaminant Conc	nnm	15	Calculations Targeted Compounds	
Mass of Contaminant - water	b	13.6	Ave Mol Mass of Targeted Compounds g/mol	92
Mass of Contaminant - water	Grame	6102.4	Molos	492.2
Mass of Contaminant - water	Granis	94.1	Mole Mass of H2O2	402.2
Mass of Contaminant -soll	D.	29472.6		40CEI
Mass of Contaminant -soli	Grams	30173.0	Mole Mass of Paraulfate	#REF!
Mass of Contaminent Targeted	Grans	44300.0		200 4
	IDS	97.7	Moles of Na25208	7009.1
Calculated soil conc.	ppm	15.1		
Ratio of S203 to targeted Compouns	Ratio	20		
Grams of sodium persultate	grams	887321.0	Molar Ratio Calc	
Pounds of Sodium Persulfate Required	Pounds	1954.5	Targeted Compounds	1
Allocation per compound (persulfate)	%	100.0%		
Total Pounds of Sodium Persulfate Required	Pounds	1954.5		
Decomposition Rate of Sodium Persulfate	%/day	1.10%		
Targeted Longevity of Persulfate	days	80		
Total Persulfate Calcualted dosage	pounds	3674		
Total Pounds of Ferric Oxide Required	Pounds	367		
Injection Summary				
Number of Injection Locations		4		
Injection Depth		20-35'		
Pounds of Persulfate		3700		
Pounds of Ferric Oxide		370		
Pounds of Persulfate/Ferric Oxide		4070		
Minimum Pounds of MgO				
Actual Pounds of MgO		0		
Injection Point Summary - Number of Intervals		2		
Injection Zones		20-25', 30-35'		
Pounds of Sodium Persulfate w / longevity	per well	463	Gallons of Water needed (area)	2400
Pounds of Ferric Oxide	per well	46		
Pounds of Persulfate/Ferric	per well	509		
Cost Basis Summary	Units			
Pounds of Persulfate/Ferric Oxide	4070			
Number of Injection Points per Event	4			
Days of Injection Trailer (3 Man Crew)	2.75			

INJECTION WELL INSTALLATION

	Area (FtXFt)	Mass of Soil/BR (tons)	Lbs FeO/Persulfate	Gallons 50% H2O2	Lbs FeSO4	Injection Pts	Days
Treatment Areas							
IW-A	907	403.32	0	12	23	1	0.2
IW-B	907	403.32	0	18	23	1	0.2
IW-2A	907	403.32	0	26	23	1	0.2
IW-3A	907	403.32	0	26	23	1	0.2
IW-3B	908	403.32	0	26	23	1	0.2
IW-2b	908	403.32	770	0	0	1	0.25
New Wells	3,630	3,226.67	4,070	0	0	8	2.75
Total	9,076	5,646.56	4,840	107	115	14	4
				Line Items			Unit
				Well Installation			LS
				Materials Handling and	I Forklift Renta	1	1
				Hazmat Shipping			LS
				Per Diem (2 man Crew)		7
				Private Utility Markout			0.5
				Administrative and Rep	orting Cost		LS
				Total Project Cost			